

- Alerts you to adherence problems in real time through visual alarms
- Monitors the performance of specific individuals in real time
- Provides information to take immediate corrective action
- Provides at-a-glance view of agent activity
- Shows agent log-on and log-off times
- Provides alarm summary reports for specified agents over a specified span of time
- Sends pop-up messages to Real-Time Adherence users

About the Architecture

Figure 1-1 shows an overview of the Real-Time Adherence architecture. Real-Time Adherence monitors real-time event messages from your ACD; each message is associated with an employee login ID. Using this information, Real-Time Adherence continually examines each employee's ACD status and compares it to the employee's scheduled status using schedule information from Aspect eWorkforce Management. Based on the preferences you specify, Real-Time Adherence notifies you when actual and scheduled activities are not in synchronization.



Figure 1-1 Real-Time Adherence Architecture

About Alarm Windows and Workspaces

Real-Time Adherence creates *alarm windows*, which are used for tracking your employees' activities, and displays these alarm windows in a working environment called a *workspace*. Real-Time Adherence tracks your employees' ACD states and determines whether they are adhering to their assigned schedules. For example, an employee who is logged out of the ACD during a break is in adherence, while an employee who is logged in to the ACD during a break is out of adherence. (You can use Real-Time Adherence to create exceptions to this principle.)

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 $\text{H}_2\text{O}(\text{g}) + \frac{1}{2} \text{O}_2(\text{g}) \rightarrow \text{H}_2\text{O}(\text{l}) \quad \Delta H = 71.2 \text{ kJ}$
 $\text{C}_2\text{H}_6(\text{g}) + \frac{7}{2} \text{O}_2(\text{g}) \rightarrow 2 \text{ CO}_2(\text{g}) + 3 \text{ H}_2\text{O}(\text{g}) \quad \Delta H = 283 \text{ kJ}$
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 $\text{C}_2\text{H}_4(\text{g}) + 3 \text{ O}_2(\text{g}) \rightarrow 2 \text{ CO}_2(\text{g}) + 2 \text{ H}_2\text{O}(\text{l}) \quad \Delta H = 1411 \text{ kJ/mole}$
 $\text{C}_2\text{H}_6(\text{g}) + 7/2 \text{ O}_2(\text{g}) \rightarrow 2 \text{ CO}_2(\text{g}) + 3 \text{ H}_2\text{O}(\text{l}) \quad \Delta H = 1560 \text{ kJ/mole}$
 $\text{H}_2(\text{g}) + 1/2 \text{ O}_2(\text{g}) \rightarrow \text{H}_2\text{O}(\text{l}) \quad \Delta H = 285 \text{ kJ/mole}$

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