

# Experiment 17 Report Sheet

## Antacid Analysis

Date \_\_\_\_\_ Lab Sec. \_\_\_\_\_ Name \_\_\_\_\_ Desk No. \_\_\_\_\_

### A. Dissolving the Antacid

	<i>Trial 1</i>	<i>Trial 2</i>	<i>Trial 1</i>	<i>Trial 2</i>
1. Mass of flask (g)	123.9596	124.1273	124.3741	
2. Mass of flask + antacid sample (g)	124.1331	124.3271	124.5632	
3. Mass (or tared mass) of antacid sample (g)	0.1735	0.1998	0.1941	
4. Total volume of HCl added (mL)	25.00	25.00	25.00	
5. Molar concentration of HCl (mol/L)	0.0924		0.0924	

### B. Analyzing the Antacid Sample

	<i>Trial 1</i>	<i>Trial 2</i>	<i>Trial 1</i>	<i>Trial 2</i>
1. Molar concentration of NaOH (mol/L)	0.10577		0.10577	
2. Buret reading, initial (mL)	1.00	23.00	23.00	
3. Buret reading, final (mL)	14.00	35.50	35.10	
4. Volume of NaOH (mL)	13.00	12.50	12.10	

### C. Calculations

1. Moles of HCl added, total (mol)	0.00231	0.00231	0.00231	
2. Moles of NaOH added (mol)	0.00138	0.00132	0.00128	
3. Moles of base in antacid sample (mol)	0.00093	0.00099	0.00103	
4. $\frac{\text{mol base in antacid}}{\text{mass of antacid sample}}$ (mol/g)	0.00536 *	0.00495	0.00531 *	
5. Average $\frac{\text{mol base in antacid}}{\text{mass of antacid sample}}$ (mol/g)	0.00516		0.00531	

\*Show calculation(s) for Trial(s) 1 on the next page.

# Experiment 17 Laboratory Manual

**Jay L. Nadeau**



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