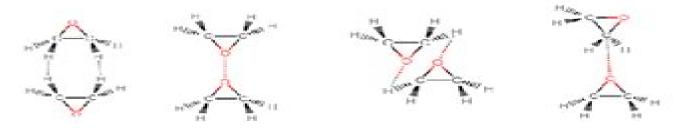
Worksheet #1 Chapter 12

Q1

How do ethylene oxide molecules interact with each other?



Q2

Which pairs of compounds will form homogenous solutions? For each pair that forms homogenous solutions, what are the intermolecular forces?

- (a) CCl₄ and H₂O
- (b) KCl and H₂O
- (c) CH₂CH₂OH and H₂O
- (d) Br₂ and CCl₄

Q3

- (a) For the pair of compounds CH₂CH₂CH₂OH or CH₃OH the one with the higher vapor pressure at a given temperature is
- (b) For the pair of compounds CH₄ or CH₃Cl the one with the higher vapor pressure at a given temperature is ______?
- (c) For the pair of compounds CH₂OH or H₂CO the one with the higher vapor pressure at a given temperature is _____?

Q4

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