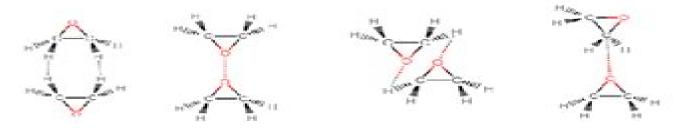
#### Worksheet #1 Chapter 12

#### Q1

How do ethylene oxide molecules interact with each other?



#### Q2

Which pairs of compounds will form homogenous solutions? For each pair that forms homogenous solutions, what are the intermolecular forces?

- (a) CCl<sub>4</sub> and H<sub>2</sub>O
- (b) KCl and H<sub>2</sub>O
- (c) CH<sub>2</sub>CH<sub>2</sub>OH and H<sub>2</sub>O
- (d) Br<sub>2</sub> and CCl<sub>4</sub>

Q3

- (a) For the pair of compounds CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>OH or CH<sub>3</sub>OH the one with the higher vapor pressure at a given temperature is
- (b) For the pair of compounds CH<sub>4</sub> or CH<sub>3</sub>Cl the one with the higher vapor pressure at a given temperature is \_\_\_\_\_\_?
- (c) For the pair of compounds CH<sub>2</sub>OH or H<sub>2</sub>CO the one with the higher vapor pressure at a given temperature is \_\_\_\_\_?

#### Q4

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**David Baker** 

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