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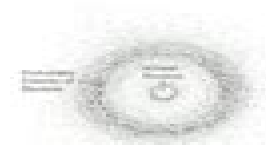
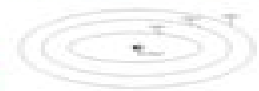
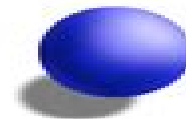
## Atomic Structure & Electron Arrangement Test Review

### Practice Questions:

1. The atomic number is equal to the number of **protons**.
2. The mass number is equal to the **protons+ neutrons**.
3. Electrons in the outermost energy level are known as **valence electrons** and are available to be lost, gained or shared when molecules are formed.
4. What causes an atom to be neutrally charged?  
**An equal number of protons and electrons**
5. Describe how an atom would become negatively charged.  
**To be negatively charged, an atom must have gained more electrons.**

Identify the major contribution of each of the following to current atomic theory (be sure to include any model that they proposed and the experiment that they did):

6. Aristotle – **everything was made from earth, wind, fire, water, and ether; alchemist**
7. Democritus – **first to propose the idea of an atom (indivisible particle)**
8. Dalton – **developed the basis for the modern atomic theory**
9. Thomson - **used the cathode ray tube to discover the first subatomic particle - the electron; developed the Plum Pudding Model**
10. Rutherford - **Gold foil experiment- expected all of the radiation to pass through and was very surprised when some of the particles were deflected - led to the discovery of the NUCLEUS (centrally located positive charge) and the concept that the atom is mostly empty space. Also discovered the proton and neutron.**
11. Bohr - **His atomic model (planetary model) had atoms built up of successive orbital shells of electrons. Electrons move between shells based on specific amounts of energy being lost or gained.**
12. Electron Cloud Model - **This model is based on quantum mechanics which states that you cannot know the speed and location of a particle. So this model shows the most probable location of the electrons.**



# Electron Structure Test Review Guide

**F Rizvi**



## **Electron Structure Test Review Guide:**

The book delves into Electron Structure Test Review Guide. Electron Structure Test Review Guide is a crucial topic that must be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Electron Structure Test Review Guide, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:

- Chapter 1: Introduction to Electron Structure Test Review Guide
- Chapter 2: Essential Elements of Electron Structure Test Review Guide
- Chapter 3: Electron Structure Test Review Guide in Everyday Life
- Chapter 4: Electron Structure Test Review Guide in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, this book will provide an overview of Electron Structure Test Review Guide. This chapter will explore what Electron Structure Test Review Guide is, why Electron Structure Test Review Guide is vital, and how to effectively learn about Electron Structure Test Review Guide.

3. In chapter 2, the author will delve into the foundational concepts of Electron Structure Test Review Guide. The second chapter will elucidate the essential principles that must be understood to grasp Electron Structure Test Review Guide in its entirety.

4. In chapter 3, this book will examine the practical applications of Electron Structure Test Review Guide in daily life. This chapter will showcase real-world examples of how Electron Structure Test Review Guide can be effectively utilized in everyday scenarios.

5. In chapter 4, this book will scrutinize the relevance of Electron Structure Test Review Guide in specific contexts. The fourth chapter will explore how Electron Structure Test Review Guide is applied in specialized fields, such as education, business, and technology.

6. In chapter 5, this book will draw a conclusion about Electron Structure Test Review Guide. The final chapter will summarize the key points that have been discussed throughout the book.

The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Electron Structure Test Review Guide.

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